# The development of a simple electrochemical method for the selective determination of caffeic acid in wine using a molecularly imprinted polymer

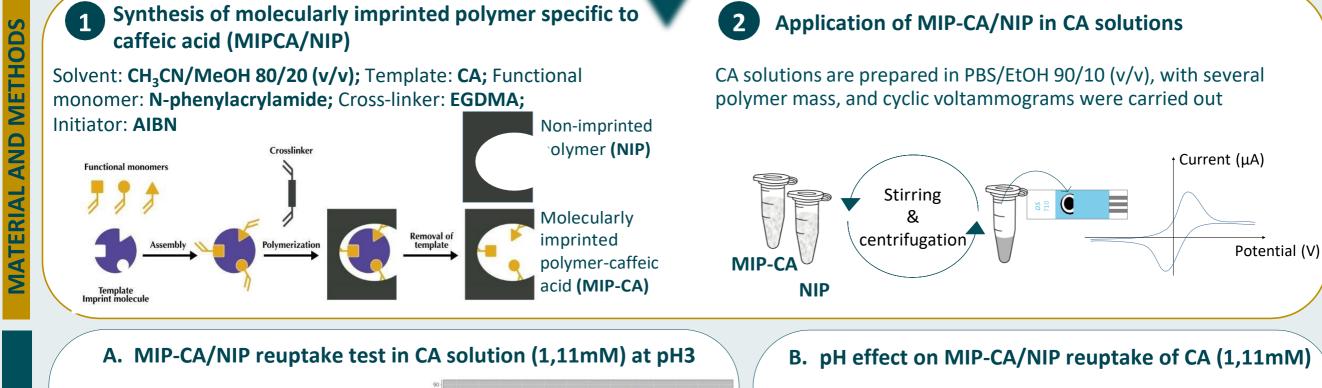
## Marie Elhachem <sup>1,2</sup>, Elias Bou-Maroun<sup>1</sup>, Maher Abboud <sup>3</sup>, Richard G. Maroun <sup>2</sup>, Philippe Cayot <sup>1</sup>

<sup>1</sup> UMR PAM Procédés Alimentaires et Microbiologiques, Université de Bourgogne Franche Comté/AgroSup Dijon, Fance <sup>2</sup> Laboratoire CTA, UR TVA, Centre d'Analyses et de Recherche, Faculté des Sciences, Université Saint Joseph de Beyrouth, Liban <sup>3</sup> UEGP Unité environnement, génomique et protéonique, Faculté des Sciences, Université Saint Joseph Beyrouth, Liban marie.el-hachem@agrosupdijon.fr

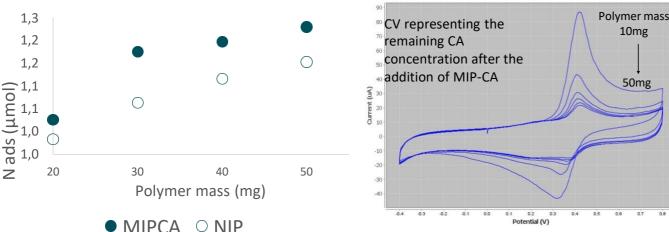
### CONTEXT

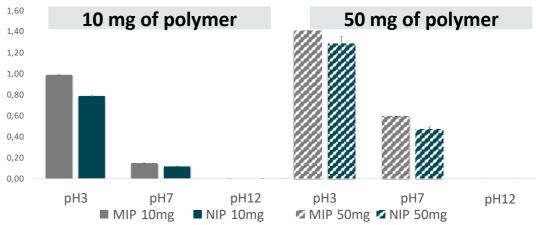
Caffeic acid (CA) is a potent antioxidant and one of the main hydroxycinnamic acids present in wine<sup>1</sup>. The use of traditional methods for its determination is costly, time consuming and require sample preparation<sup>2</sup> **OBJECTIVE OF THIS STUDY** 

Combining electrochemistry and molecular imprinting in order to obtain a simple method used for the selective determination of caffeic acid in hydroalcoholic medium, and its application in wine



N ads (µmol)





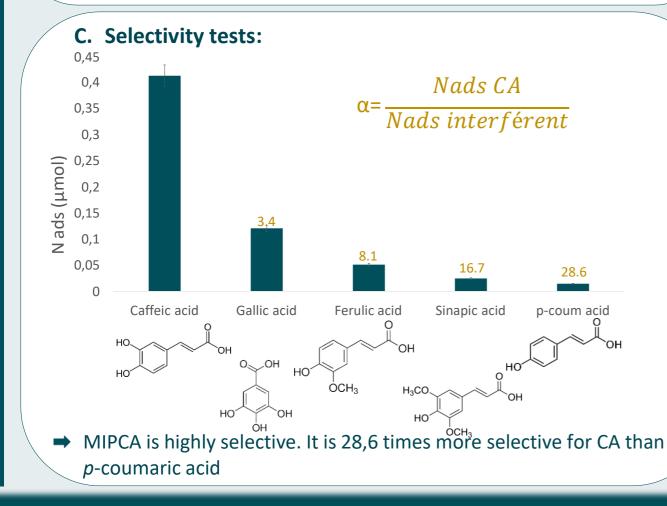
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Reuptake activity is proportional to the added polymer mass



- → The best reuptake is carried out at pH3, where the majority of caffeic acid molecules are protonated
- D. Application in wine: results of addition-recovery experiments using MIP-CA for determination of CA concentration

Wine with added CA	Added [CA] (μmol/L)	Found [CA] (µmol/L)	Recovery (%)
1 <sup>st</sup> level	110	140	120
2 <sup>nd</sup> level	240	230	100
3 <sup>rd</sup> level	660	540	80

MIPCA showed a reuptake activity of CA in wine, with recoveries ranging between 80% and 120%, indicating that the matrix had no influence on the polymer

#### **CONCLUSIONS AND PERSPECTIVES**

- > This method allowed a rapid and easy determination of caffeic acid. Electrochemistry requires a very small volume of sample without any preparation and the preparation of the polymer is relatively simple.
- > MIPCA is highly selective for caffeic acid in model and real wine.

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> It would be interesting to test the same MIPCA on MWCNT electrodes, and compare the signals.

<sup>1</sup>: Ilhami G. (2006). *Toxicology*.;217:213-220; <sup>2</sup>: Elhachem, M. et al. (2021). *Antioxidants*.;10(3).

